

Impacts of the production and consumption of biofuels on stratospheric ozone

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[1] Biofuels are becoming increasingly popular sources of renewable energy as economic pressures and environmental consequences encourage the use of alternatives to fossil fuels. However, growing crops destined for use as biofuels incurs large N₂O emissions associated with the use of nitrogen-based fertilizers. Besides being a greenhouse gas, N₂O is also the primary source of stratospheric NO_x (NO + NO₂) which leads to stratospheric ozone depletion. In this paper, the potential effects on the ozone layer of a large-scale shift away from fossil fuel use to biofuels consumption over the 21st century are examined. Under such a scenario, global-mean column ozone decreases by 2.6 DU between 2010 and 2100 in contrast to a 0.7 DU decrease under a control simulation (the IPCC SRES B1 scenario for greenhouse gases) and a 9.1 DU increase under the more commonly used SRES A1B scenario. Two factors cause the decrease in ozone in the biofuels simulation: 1) large N₂O emissions lead to faster rates of the ozone-depleting NO_x cycles and; 2) reduced CO₂ emissions (due to less fossil fuel burning) lead to relatively less stratospheric cooling over the 21st century, which decreases ozone abundances. Reducing CO₂ emissions while neglecting to reduce N₂O emissions could therefore be damaging to the ozone layer. **Citation:** Revell, L. E., G. E. Bodeker, P. E. Huck, and B. E. Williamson (2012), Impacts of the production and consumption of biofuels on stratospheric ozone, *Geophys. Res. Lett.*, 39, L10804, doi:10.1029/2012GL051546.

1. Introduction

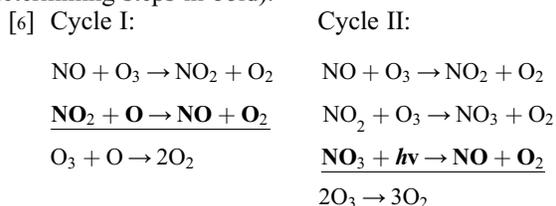
[2] As fossil fuel reserves diminish and awareness of the detrimental effects of fossil fuel-burning on global climate continues to grow, biofuels are becoming an increasingly attractive supply of fuel, especially in Europe, the USA and Brazil [Bessou *et al.*, 2011]. In 2010, biofuels accounted for 2.7% of road-transportation fuel use globally (<http://vitalsigns.worldwatch.org/vs-trend/biofuels-regain-momentum>). Biofuels are often considered to be carbon neutral because the CO₂ released on burning had previously been removed from the atmosphere via photosynthesis. However, nitrogen-based fertilizers used in growing the crops from which biofuels are produced lead to N₂O emissions via soil nitrification and

denitrification [Smeets *et al.*, 2009]. N₂O is a greenhouse gas (GHG) with a 100-year global warming potential of ~298, and a lifetime of ~114 years [Forster *et al.*, 2007].

[3] Crutzen *et al.* [2008] suggested that the global warming reduction achieved through CO₂ reductions by using first-generation biofuels (those produced from vegetable oil, starch or sugar) instead of fossil fuels, could be countered by the subsequent increase in N₂O emissions resulting from increased nitrogen-based fertilizer use.

[4] Using linked economic and terrestrial biogeochemistry models, Melillo *et al.* [2009] found that carbon emissions resulting from land-use change under a global biofuels programme would be significant. However, they also found that an increase in N₂O emissions due to increases in nitrogen-based fertilizer use would be a more important contributor to climate warming. Additionally, under two different scenarios in which biofuels production was projected to increase, they estimated this would account for ~60% of total annual N₂O emissions by 2100.

[5] As well as being a GHG, N₂O leads to stratospheric ozone destruction through its reaction with O(¹D) to produce NO_x (NO + NO₂) [Crutzen, 1970]. NO_x depletes ozone by participating catalytically in the following cycles (rate-determining steps in bold):



[7] N₂O is projected to be the dominant ozone-depleting gas emitted through the 21st century [Ravishankara *et al.*, 2009]. However, stratospheric cooling resulting from increases in other GHGs, primarily CO₂, decreases the effectiveness of N₂O as an ozone-depleting substance by: 1) increasing the sink for reservoir nitrogen species [Rosenfield and Douglass, 1998], and 2) decreasing the O/O₃ ratio [Jonsson *et al.*, 2004], thus slowing the NO₂ + O rate-determining step of Cycle I by decreasing the abundance of atomic oxygen. Furthermore, as sea-surface temperatures (SSTs) increase over the 21st century, the Brewer-Dobson circulation (BDC) is projected to strengthen [Fomichev *et al.*, 2007], thus reducing the yield of NO_x from N₂O [Cook and Roscoe, 2009].

[8] Three chemistry-climate model (CCM) simulations – ‘A1B’, ‘B1’ and ‘biofuels’ (Table 1) have been performed to examine the potential effects of increased production and consumption of biofuels on stratospheric ozone. The A1B and B1 simulations were based on the IPCC SRES A1B and B1 GHG emissions scenarios, respectively [Nakicenovic

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Table 1. Summary of the Three CCM Simulations

	A1B	B1	Biofuels
CO ₂ in 2100 (GtC) ^a	13.1	5.2	3.7
N ₂ O in 2100 (MtN ₂ O-N) ^a	7.0	5.7	14.3
2090s temperature (K) ^b	215.2	218.6	218.9
ΔO ₃ (DU) ^c	9.1	-0.7	-2.6

^aCO₂ and N₂O emissions.

^bGlobal-mean temperatures at 30 hPa simulated by NIWA-SOCOL, and averaged over the 2090s decade.

^cChange in global-mean total column ozone over the 21st century simulated by NIWA-SOCOL (2090s decade minus 2010s decade), expressed in Dobson Units.

and Swart, 2000]. The SRES A1B scenario portrays an intermediate increase in concentrations of anthropogenic GHGs, while SRES B1 prescribes relatively low GHG emissions. The A1B and B1 simulations display similar N₂O surface concentrations but different CO₂ surface concentrations and SSTs, which influence stratospheric temperatures and the BDC. This facilitated an investigation of how the BDC and CO₂-induced stratospheric cooling influence ozone and ozone-depleting NO_x chemistry. The major difference between the B1 and biofuels simulations was the N₂O surface concentrations, permitting a study of the effect of increased N₂O emissions on ozone as a result of increased nitrogen-based fertilizer use.

2. Computational Methods

[9] CCM simulations were performed using the NIWA-SOCOL (National Institute of Water and Atmospheric Research – Solar Climate Ozone Links) model. NIWA-SOCOL is based on version 2.0 of the SOCOL model [Schraner *et al.*, 2008; SPARC CCMVal, 2010], and includes 41 chemical species, 140 gas-phase reactions, 46 photolysis reactions and 16 heterogeneous reactions. The reaction rate constants used were those recommended by Sander *et al.* [2006, 2009]. NIWA-SOCOL performs chemical and radiative calculations every two hours, and dynamical process calculations every 15 minutes. The adjusted A1 scenario for halocarbon concentrations that accounts for the accelerated phase-out of HCFCs was used [Daniel *et al.*, 2007], while the GHG concentrations prescribed for the three simulations are described in more detail below.

[10] NIWA-SOCOL tracks the contribution to ozone loss from 15 catalytic cycles, similar to the diagnostic approach used by Lee *et al.* [2002]. Odd oxygen (O + O(¹D) + O₃) loss rates (in molecules cm⁻³ s⁻¹) were calculated within the model using the rate-limiting steps of the corresponding reaction cycles. Results were recorded and accumulated into daily means within each model grid cell.

[11] A simple climate model, MAGICC6 (Model for the Assessment of Greenhouse-gas Induced Climate Change version 6) [Meinshausen *et al.*, 2011], was used to convert the emissions scenarios to the surface concentration scenarios required as input to the CCM simulations. Prescribing concentrations rather than emissions boundary conditions breaks the feedback of ozone changes on changes in tropospheric lifetimes. However for our study, omitting this feedback does not affect our conclusions. MAGICC6 can be calibrated to emulate any of the 19 CMIP3 (Coupled Model Intercomparison Project phase 3) Atmosphere-ocean General

Circulation Model (AOGCMs) and the 10 C⁴MIP (Coupled Carbon Cycle Climate Model Intercomparison Project) carbon cycle models. For our calculations, MAGICC6 was calibrated to the ECHAM5/MPIOM AOGCM core model parameters, and LLNL model carbon cycle parameters.

3. Emissions Scenarios

[12] For the B1 simulation, surface concentrations of CO₂, N₂O and CH₄, commensurate with the SRES B1 emissions scenario, were calculated using MAGICC6 tuned as described above. SSTs and sea-ice concentrations (SICs) were prescribed using output from the UKMO-HadCM3 AOGCM, based on the SRES B1 emissions scenario [Nakicenovic and Swart, 2000]. The SST and SIC datasets were obtained from the World Climate Research Programme's CMIP3 multi-model dataset.

[13] Since the biofuels simulation was constructed to follow the same radiative forcing pathway as the B1 simulation (see below), it used the same SSTs, SICs and CH₄ surface concentrations as the B1 simulation. The biofuels simulation was based on findings from Melillo *et al.* [2009], which suggest that biofuels production and the associated use of nitrogen-based fertilizers could account for ~60% of total annual N₂O emissions by 2100. To capture this impact, an emissions scenario for N₂O, with emissions of 14.3 MtN₂O-N in 2100 compared with 5.7 MtN₂O-N in the B1 scenario, was constructed. MAGICC6 calculated the corresponding surface N₂O concentrations, which were 443 ppb in 2100 compared with 368 ppb in the B1 scenario (Figure 1a).

[14] CO₂ emissions in the biofuels scenario were decreased relative to those prescribed in the B1 scenario such that the global-mean radiative forcing in the biofuels simulation was the same as in the B1 simulation. By keeping the radiative forcing the same in both simulations, we can ensure that any differences seen are attributable to changes in chemistry and are not potentially compromised by changes in dynamics. The reduction in CO₂ concentrations also served to simulate the expected reduction in CO₂ emissions resulting from biofuels use rather than fossil fuels use. This resulted in CO₂ emissions of 3.7 GtC in 2100 in the biofuels scenario, compared with 5.2 GtC in the B1 scenario (summarized in Table 1, along with N₂O emissions). The equivalent CO₂ surface concentrations in 2100, calculated using MAGICC, were 532 ppm in the biofuels scenario, compared with 551 ppm in the B1 scenario (Figure 1b).

[15] The A1B simulation used GHG concentrations, SSTs and SICs consistent with the SRES A1B scenario. SSTs and SICs were taken from ECHAM5/MPIOM AOGCM output. SSTs for the A1B and B1 simulations are shown in Figure 1d. CH₄ concentrations for the three scenarios are displayed in Figure 1c.

4. Results and Discussion

[16] The projected changes in global-mean total column ozone over the 21st century for the three simulations are summarized in Table 1. In our A1B simulation, global-mean total column ozone is projected to increase by 9.1 DU through the 21st century, due to a slowing of the halogen ozone-loss cycles (as concentrations of stratospheric

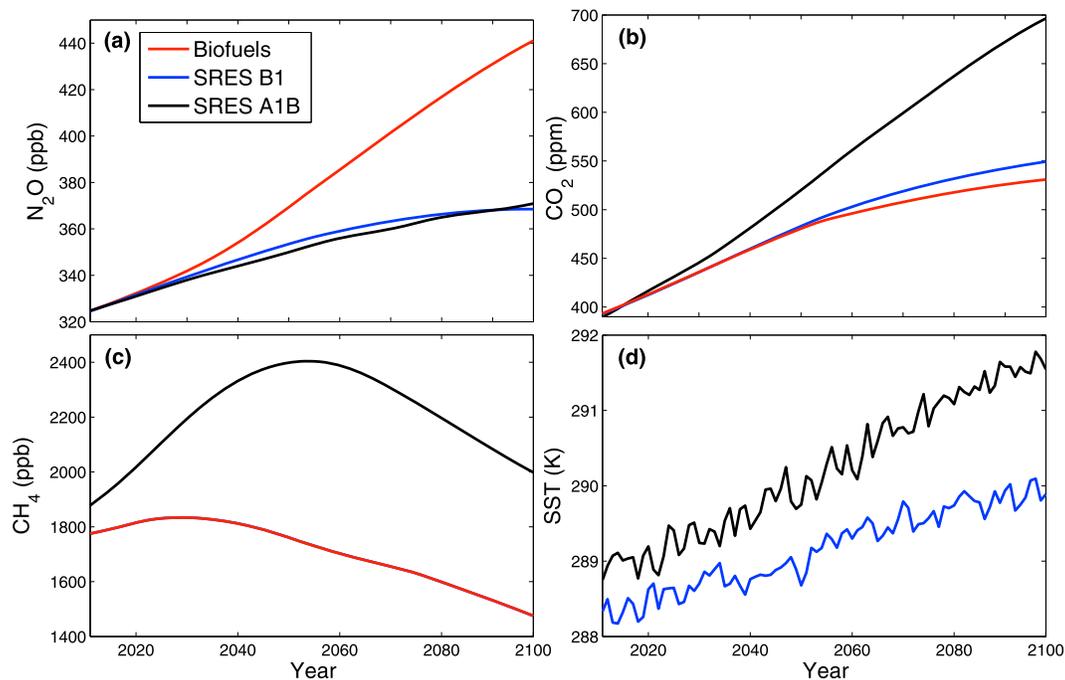


Figure 1. (a) N_2O , (b) CO_2 , (c) CH_4 surface concentrations for the A1B, B1 and biofuels emissions scenarios. CH_4 concentrations are identical for the B1 and biofuels scenario such that the red and blue traces overlay each other. (d) Global-mean SSTs for the A1B and B1 simulations (the biofuels simulation used B1 SSTs).

chlorine and bromine decline) and GHG-induced stratospheric cooling (which slows the temperature-dependent ozone loss cycles [Rosenfield *et al.*, 2002]). This result is consistent with other CCM projections in the SPARC (Stratospheric Processes and their Role in Climate) CCM Validation (CCMVal-2) assessment [SPARC CCMVal, 2010], which also used the SRES A1B scenario.

[17] In the B1 simulation, column ozone decreases by 0.7 DU over the 21st century since stratospheric cooling (which induces ozone increases) is relatively less than in the A1B simulation where larger stratospheric cooling leads to a 9.1 DU increase in ozone. A larger decrease in column ozone of 2.6 DU is calculated in the biofuels simulation, because greater N_2O emissions lead to enhanced rates of the ozone-depleting NO_x cycles.

[18] Ozone differences between the B1 and A1B simulations at the end of the 21st century are displayed in Figure 2a as a percentage of ozone in the A1B simulation, and in Figure 2b as the difference in total column ozone (B1 minus A1B). This facilitates an assessment of the effect of GHG forcing (for example, driving changes in stratospheric temperature and the BDC) on ozone. The ozone concentrations in the B1 simulation in the upper and lower Antarctic stratosphere and the tropical lower stratosphere are greater than those in the A1B simulation, consistent with the findings of Eyring *et al.* [2010]. Elsewhere, the B1 simulation ozone concentrations are lower by as much as 20%.

[19] The lower ozone concentrations in the B1 simulation are likely due to: 1) the weaker stratospheric cooling, with the result that the temperature-dependent gas-phase ozone loss cycles slow less than in the A1B simulation; 2) lower CH_4 concentrations than in the A1B scenario [Portmann and Solomon, 2007; Fleming *et al.*, 2011] and; 3) reduced strengthening of the BDC, as a result of cooler SSTs

[Fomichev *et al.*, 2007; Bekki *et al.*, 2011], which means that the ozone flux into the southern mid-latitude lower stratosphere is weaker than in the A1B simulation.

[20] Increased tropical lower stratospheric ozone in the B1 simulation is consistent with a difference in the BDC between the B1 and A1B simulations. The slower ascent rate in the tropical lower stratosphere in the B1 simulation allows more time for ozone to form in the rising air parcels than in more quickly ascending air in the A1B simulation [Avallone and Prather, 1996]. Elevated lower-stratospheric Antarctic ozone in the B1 simulation is caused by a relatively warmer lower stratosphere; over the 21st century, Antarctic winter temperatures at 50 hPa average 197 K in the B1 simulation compared to 189 K in the A1B simulation. As a result, polar stratospheric cloud formation, and the associated heterogeneous chlorine and bromine ozone-depleting chemistry, is suppressed in the B1 simulation compared to the A1B simulation. In the 2090s, there is $\sim 5\%$ more upper-stratospheric Antarctic ozone in the B1 simulation than in the A1B simulation because there is less CH_4 , and hence the ozone-depleting HO_x cycles (which are dominant in the upper stratosphere) are slower.

[21] Figure 2c displays the same quantity as Figure 2a, but calculated for the difference between the biofuels and B1 simulations. This facilitates an assessment of the effects of larger N_2O emissions on ozone. Figure 2c shows that under the biofuels scenario, ozone concentrations are (relatively) suppressed throughout the middle stratosphere and enhanced in the lower stratosphere. However, because the middle stratosphere dominates the ozone column, total column amounts are smaller in the biofuels simulation and up to 11 DU smaller at northern high latitudes (Figure 2d).

[22] In the biofuels simulation, ozone is greater by up to 4% in the troposphere compared to that in the B1 simulation.

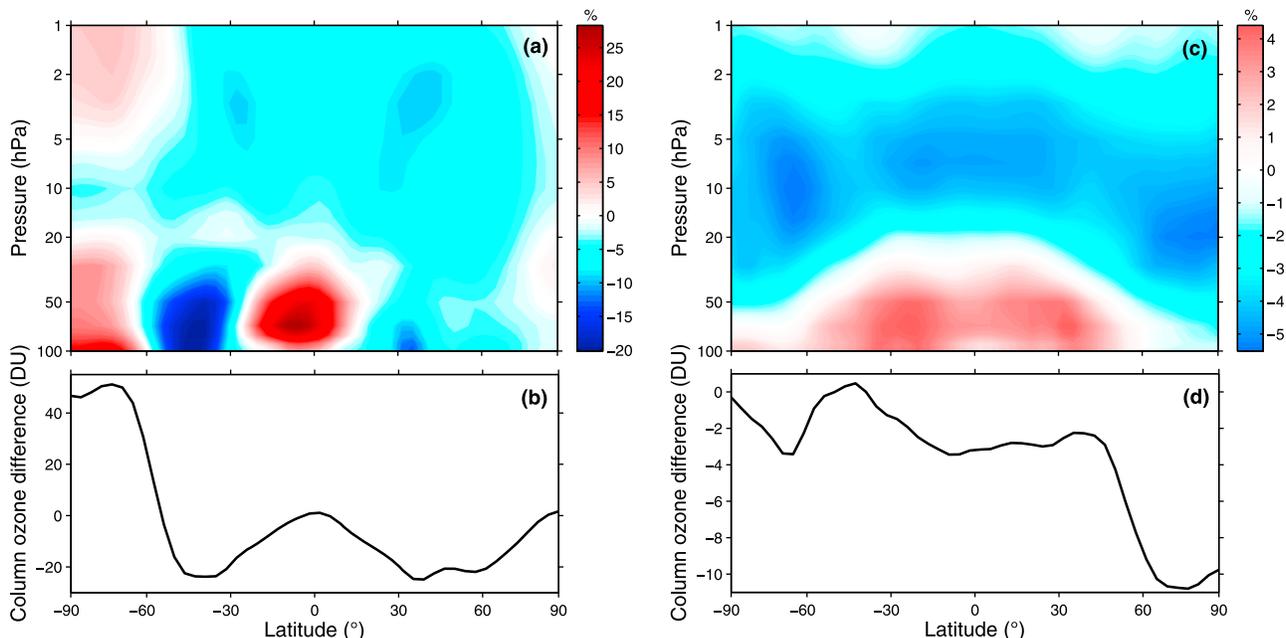
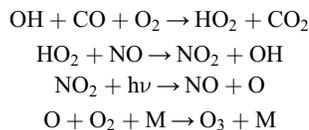


Figure 2. (a) B1 ozone minus A1B ozone in the 2090s decade, calculated as a percentage of ozone in the A1B simulation. (b) 2090s decade B1 total column ozone minus A1B total column ozone. (c) Biofuels ozone minus B1 ozone in the 2090s decade, calculated as a percentage of ozone in the B1 simulation. (d) 2090s decade biofuels total column ozone minus B1 total column ozone.

N_2O leads to an increase in tropospheric ozone production, via the following reactions involving NO_x :



Ozone production by this mechanism is generally insignificant in the stratosphere, where the concentration of CO is too small [Lanzendorf *et al.*, 2001].

[23] In the middle and upper stratosphere, ozone in the biofuels scenario is up to 5% less than that in the B1 simulation, since N_2O concentrations are higher, leading to a faster rate of the ozone-depleting NO_x cycles. These cycles

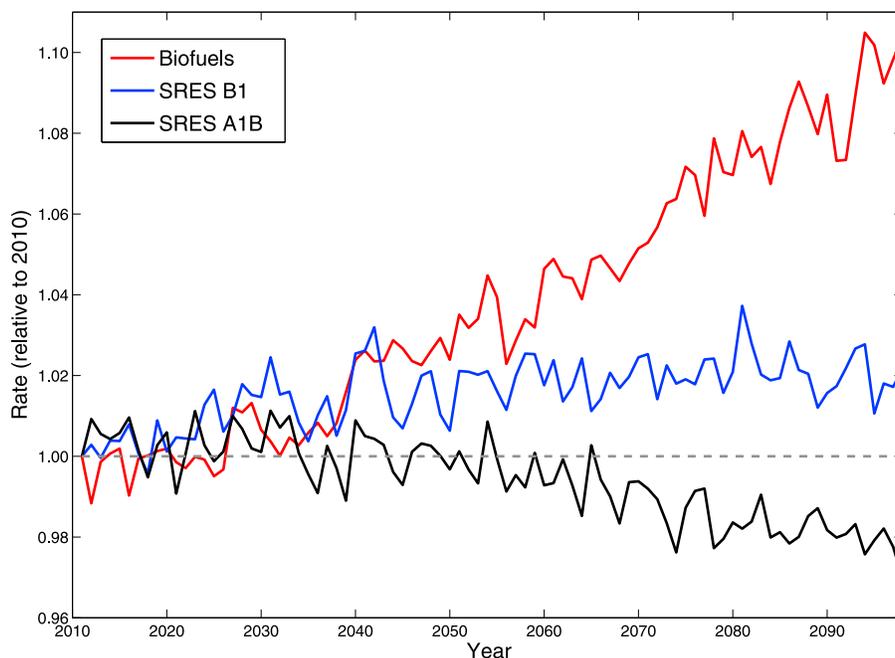


Figure 3. Global-mean rate of the NO_x ozone loss cycles, averaged over 1–100 hPa and normalized to 1.0 in 2010. The differences between the biofuels and B1 simulations are largely a consequence of different N_2O emissions, while those between the B1 and A1B simulations are largely due to temperature differences.

do not occur in the troposphere where O and O₃ concentrations are minimal.

[24] Figure 3 shows the combined rate of the two NO_x ozone-loss cycles (Cycles I and II), averaged over 1–100 hPa, relative to their rate in 2010 for each of the three simulations. Despite increasing N₂O emissions in the A1B simulation, NO_x-induced ozone depletion slows over the 21st century (as described by L. E. Revell et al., The effectiveness of N₂O in depleting stratospheric ozone, submitted to *Geophysical Research Letters*, 2011). The B1 simulation has N₂O emissions almost identical to those in the A1B simulation, yet because the stratosphere cools relatively less, the rate at which the NO_x cycles deplete ozone increases by ~2% between 2010 and 2100. Finally, in the biofuels simulation, the rate of ozone-depleting NO_x chemistry increases by ~10% over the 21st century owing to the large increase in N₂O concentrations.

5. Conclusions

[25] We have presented the effects of three different GHG emissions scenarios on stratospheric ozone. The SRES A1B scenario is the scenario on which the majority of the SPARC CCMVal-2 simulations were based. In our A1B simulation, global-mean total column ozone increased by 9.1 DU over the 21st century due to decreasing halogen concentrations and GHG-induced stratospheric cooling. Such an increase was not observed in the simulation based on the SRES B1 scenario (0.7 DU decrease), which has lower CO₂ and CH₄ emissions. Increasing N₂O emissions increase the rate of NO_x-catalyzed ozone loss, and this effect becomes more pronounced when the stratosphere cools relatively less, as it does in the biofuels simulation (2.6 DU decrease). Therefore, the reduction in CO₂ emissions achieved by switching to biofuels is not large enough to compensate for the deleterious effects of associated increases in N₂O emissions on stratospheric ozone. Increased biofuels production and consumption could therefore be damaging to the ozone layer.

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References

- Avallone, L. M., and M. J. Prather (1996), Photochemical evolution of ozone in the lower tropical stratosphere, *J. Geophys. Res.*, *101*(D1), 1457–1461, doi:10.1029/95JD03010.
- Bekki, S., et al. (2011), Future ozone and its impact on surface UV, in *Scientific Assessment of Ozone Depletion: 2010, Global Ozone Res. Monit. Proj. Rep. 52*, chap. 3, pp. 1–60, World Meteorol. Organ., Geneva, Switzerland.
- Bessou, C., F. Ferchaud, B. Gabrielle, and B. Mary (2011), Biofuels, greenhouse gases and climate change. A review, *Agron. Sustainable Dev.*, *31*, 1–79, doi:10.1051/agro/2009039.
- Cook, P. A., and H. K. Roscoe (2009), Variability and trends in stratospheric NO₂ in Antarctic summer, and implications for stratospheric NO_y, *Atmos. Chem. Phys.*, *9*, 3601–3612, doi:10.5194/acp-9-3601-2009.
- Crutzen, P. J. (1970), The influence of nitrogen oxides on the atmospheric ozone content, *Q. J. R. Meteorol. Soc.*, *96*, 320–325, doi:10.1002/qj.49709640815.
- Crutzen, P. J., A. R. Mosier, K. A. Smith, and W. Winiwarter (2008), N₂O release from agro-biofuel production negates global warming reduction by replacing fossil fuels, *Atmos. Chem. Phys.*, *8*, 389–395, doi:10.5194/acp-8-389-2008.
- Daniel, J. S., et al. (2007), Halocarbon scenarios, ozone depletion potentials, and global warming potentials, in *Scientific Assessment of Ozone Depletion: 2006, Global Ozone Res. Monit. Proj. Rep. 50*, chap. 8, pp. 1–39, World Meteorol. Organ., Geneva, Switzerland.
- Eyring, V., et al. (2010), Sensitivity of 21st century stratospheric ozone to greenhouse gas scenarios, *Geophys. Res. Lett.*, *37*, L16807, doi:10.1029/2010GL044443.
- Fleming, E. L., C. H. Jackman, R. S. Stolarski, and A. R. Douglass (2011), A model study of the impact of source gas changes on the stratosphere for 1850–2100, *Atmos. Chem. Phys.*, *11*, 8515–8541, doi:10.5194/acp-11-8515-2011.
- Fomichev, V. I., A. I. Jonsson, J. de Grandpré, S. R. Beagley, C. McLandress, K. Semeniuk, and T. G. Shepherd (2007), Response of the middle atmosphere to CO₂ doubling: Results from the Canadian Middle Atmosphere Model, *J. Clim.*, *20*, 1121–1144, doi:10.1175/JCLI4030.1.
- Forster, P., et al. (2007), Changes in atmospheric constituents and in radiative forcing, in *Climate Change 2007: The Physical Science Basis. Contribution of Working Group I to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change*, edited by S. Solomon et al., pp. 129–234, Cambridge Univ. Press, Cambridge, U. K.
- Jonsson, A. I., J. de Grandpré, V. I. Fomichev, J. C. McConnell, and S. R. Beagley (2004), Doubled CO₂-induced cooling in the middle atmosphere: Photochemical analysis of the ozone radiative feedback, *J. Geophys. Res.*, *109*, D24103, doi:10.1029/2004JD005093.
- Lanzendorf, E. J., T. F. Hanisco, P. O. Wennberg, R. C. Cohen, R. M. Stimpfle, J. G. Anderson, R. S. Gao, J. J. Margitan, and T. P. Bui (2001), Establishing the dependence of [HO₂]/[OH] on temperature, halogen loading, O₃, and NO_x based on in situ measurements from the NASA-ER2, *J. Phys. Chem. A*, *105*, 1535–1542, doi:10.1021/jp002384i.
- Lee, A. M., R. L. Jones, I. Kilbane-Dawe, and J. A. Pyle (2002), Diagnosing ozone loss in the extratropical lower stratosphere, *J. Geophys. Res.*, *107*(D11), 4110, doi:10.1029/2001JD000538.
- Meinshausen, M., S. C. B. Raper, and T. M. L. Wigley (2011), Emulating coupled atmosphere-ocean and carbon cycle models with a simpler model, MAGICC6 – Part 1: Model description and calibration, *Atmos. Chem. Phys.*, *11*, 1417–1456, doi:10.5194/acp-11-1417-2011.
- Melillo, J. M., J. M. Reilly, D. W. Kicklighter, A. C. Gurgel, T. W. Cronin, S. Paltsev, B. S. Felzer, X. Wang, A. P. Sokolov, and C. A. Schlosser (2009), Indirect emissions from biofuels: How important?, *Science*, *326*, 1397–1399, doi:10.1126/science.1180251.
- Nakicenovic, N., and R. Swart (Eds.) (2000), *IPCC Special Report on Emissions Scenarios*, Cambridge Univ. Press, Cambridge, U. K.
- Portmann, R. W., and S. Solomon (2007), Indirect radiative forcing of the ozone layer during the 21st century, *Geophys. Res. Lett.*, *34*, L02813, doi:10.1029/2006GL028252.
- Ravishankara, A. R., J. S. Daniel, and R. W. Portmann (2009), Nitrous Oxide (N₂O): The dominant ozone-depleting substance emitted in the 21st century, *Science*, *326*, 123–125, doi:10.1126/science.1176985.
- Rosenfield, J. E., and A. R. Douglass (1998), Doubled CO₂ effects on NO_y in a coupled 2D model, *Geophys. Res. Lett.*, *25*, 4381–4384, doi:10.1029/1998GL900147.
- Rosenfield, J. E., A. R. Douglass, and D. B. Considine (2002), The impact of increasing carbon dioxide on ozone recovery, *J. Geophys. Res.*, *107*, 4049, doi:10.1029/2001JD000824.
- Sander, S. P., et al. (2006), Chemical kinetics and photochemical data for use in atmospheric studies, *JPL Publ.*, *06-2*, 523 pp.
- Sander, S. P., et al. (2009), Chemical kinetics and photochemical data for use in atmospheric studies: Update to key reactions, *JPL Publ.*, *09-31*.
- Schraner, M., et al. (2008), Chemistry-climate model SOCOL: Version 2.0 with improved transport and chemistry/microphysics schemes, *Atmos. Chem. Phys.*, *8*, 5957–5974, doi:10.5194/acp-8-5957-2008.
- Smeets, E. M. W., L. F. Bouwman, E. Stehfest, D. P. van Vuuren, and A. Postuma (2009), Contribution of N₂O to the greenhouse gas balance of first-generation biofuels, *Global Change Biol.*, *15*, 1–23, doi:10.1111/j.1365-2486.2008.01704.x.
- SPARC CCMVal (2010), SPARC report on the evaluation of chemistry-climate models, *SPARC Rep. 5, WCRP-132, WMO/TD 1526*, edited by V. Eyring, T. G. Shepherd, and D. W. Waugh, World Clim. Res. Programme, Geneva, Switzerland.